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A STUDY OF THE EPR SPECTRUM OF THE Ca(OH) 2: Mn²⁺

SYSTEM AS A FUNCTION OF TEMPERATURE

by

Stuart M. Quick

A Dissertation Submitted to the Faculty of Graduate Studies through the Department of Physics in Partial Fulfillment of the Requirements for the Degree of Doctor of Philosophy at the University of Windsor

Windsor, Ontario

1968

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ABSTRACT

The electron paramagnetic resonance (EPR) spectrum of Mn^{2^+} impurity in synthetic crystals of Ca(OH)₂ was studied as a function of temperature in the range $85^{\circ}K-800^{\circ}K$ at x-band microwave frequencies. The spectrum was fitted to the following spin-Hamiltonian which is invariant in D_{3d} symmetry.

$$H(s) = \beta \vec{H} \cdot \vec{g} \cdot \vec{S} + \vec{1} \cdot \vec{A} \cdot \vec{S} + \frac{1}{3} b_2^0 c_2^0 + \frac{1}{60} b_4^0 c_4^0 + \frac{1}{60} b_4^3 c_4^3$$

At RT(290°K) the following values were obtained for the parameters:

 $g_{11} = 2.0011 \pm 0.0005 \qquad g_{1} = 2.0010 \pm 0.0005 \qquad b_{2}^{0} = -6.7 \pm 0.5G$ $b_{4}^{0} = -2.48 \pm 0.25G \qquad b_{4}^{3} = 0 \pm 100G$ $A = -92.05 \pm 0.15G \qquad B = -90.28 \pm 0.15G$

The value of $b_2^0(0) = -14.25G$ which was predicted by the static theory of Sharma, Das, and Orbach, 33,34 was in good agreement with the value $b_2^0(0) = -16.25G$ which was extrapolated from the experimental data. The temperature dependence of the experimental b_2^0 , however, differed markedly from the dependence predicted by this theory when the effect of thermal expansion was taken into account. The following curve was fitted to the data:

 $b_{2}^{0}(T) = (b_{2}^{0})_{0}^{+} 4.97 \times 10^{-42} k_{1} \coth(\frac{196}{T})^{+} 11.3 \times 10^{-42} K_{2} \coth(\frac{172}{T})$ where $(b_{2}^{0})_{0} = -21.5 \pm 0.5G$, $K_{1} = (-5.38 \pm 0.5) \times 10^{42} G/erg - \sec^{2}$, and $K_{2} = (2.83 \pm 0.5)$ $\times 10^{42} G/erg - \sec^{2}$. This curve was obtained on the basis of the theory of Walsh, Jeener and Bloembergen³⁷ by assuming that two T' lattice modes transforming as A_{1g} and E_{g} under D_{3d} symmetry were responsible for the Mn²⁺

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vibronic coupling. The possibility of explaining $b_2^0(T)$ on the basis of a Debye frequency spectrum is also pointed out in the text.

The temperature dependence of b_4^0 was approximately linear. Therefore, the effects of lattice vibrations on this quantity were probably negligible. Although the data tended to support a dependence, such as $b_4^0 = CR_0^{-n}$, where C is a constant, R_0 is the Ca-O bond distance, and n is a number, it proved impossible to determine n accurately.

The theory of Simanek and Orbach⁵⁰ was used to describe the temperature dependence of the isotropic Fermi contact hyperfine tensor. A curve of the form

 $A(T) = A(0) [1 - CT⁴ \int \frac{x^3 dx}{e^{x} - 1}]$

was fitted to the data. The best fit was obtained for A(0) = -92.00G, $C = 1.78 \times 10^{-12} {}^{\circ}K^{-4}$, and a Debye Temperature of $\theta = 500^{\circ}K$. The quantity $\eta = \left| \frac{(A^2 - B^2)^{1/2}}{A} \right|$ decreased at a rate of the order of 7.6 x $10^{-50}K^{-1}$. Mn^{2^+} -H SHFS was observed in the range $85^{\circ}K$ - $130^{\circ}K$.

The interaction was highly isotropic. The isotropic Fermi contact HFS tensor was of the order of 4.5G at 85° K. The anisotropic dipolar HFS tensor was probably of the order of 0.1G or less. A dependence of the SHFS on M_s was also noted, but no theoretical explanation was given.

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CHAPTER I

INTRODUCTION AND PURPOSE OF THE EXPERIMENT

Iron group ions with d^5 configurations, e.g., Mn^{2+} and Fe^{3+} , and rare-earth group ions with f^7 configurations, e.g., Gd^{3+} and Eu^{2+} , have S ground states in Russell-Saunders coupling. Any degeneracy remaining in the state is due to spin only. It has been observed that this spin degeneracy is partially removed when the ions are present in a crystalline environment. In recent years there has been much interest in explaining such S-state splittings theoretically.

Most of the mechanisms which have been proposed to explain Sstate splittings are based on a rigid lattice model. These calculations reveal that the total S-state splitting can be the result of so-called crystal field interactions, hyperfine interactions, and sometimes, superhyperfine interactions. These interactions are discussed in detail in Chapter III. In order to determine the contribution of each of these interactions to the total splitting, one computes the magnitude of the corresponding tensor (sometimes called parameter) in the spin-Hamiltonian. The spin-Hamiltonian for the $Ca(OH)_2:Mn^{2+}$ system is introduced also in Chapter III. Walsh et. al.³⁷, in 1965, studied experimentally the temperature dependence of the so-called axial crystal field parameter (D), the cubic crystal field parameter (a), and the hyperfine parameter (A), of paramagnetic ions in cubic crystals, such as MgO. These workers found that the temperature dependences of D and a could be explained by the

1

effects of thermal expansion on a mechanism based on the rigid lattice model. The temperature dependence of A, however, could not be so explained. They proposed that the temperature dependence of A was determined mainly by the effects of lattice vibrations, and that this dependence could have been computed explicitly had the phonon spectrum of the crystal been known. Simanek and Orbach⁵⁰, in 1966, presented quantitative calculations of a mechanism which successfully describes the temperature dependence of the hyperfine parameter A of Mn^{2+} in cubic crystals.

The primary purpose of this experiment was to extend experimental studies of the type discussed above to Mn^{2^+} in trigonal symmetry, in an environment only slightly less symmetrical than cubic. We chose to study the Ca(OH)₂:Mn²⁺ system because it has a simple structure of the desired symmetry and because we were successful in introducing Mn²⁺ into the alkaline surroundings. The third reason was that the phonon spectrum of this crystal has been extensively studied. It was our intention to use such data to study the effect of Ca-OH vibrations on the temperature dependences of the spin-Hamiltonian parameters and to seek evidence of effects of internal OH vibrations on the superhyperfine structure at low temperatures.

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CHAPTER II

PHYSICAL PROPERTIES OF Ca(OH) 2 CRYSTAL

I. CRYSTALLOGRAPHY

X-ray studies¹⁻⁹ have shown Ca(OH)₂ to be of the hexagonal (CdI₂) type with space group P $\overline{3}$ 2/m 1 (D³_{3d}). A projection of the closepacked structure established by the X-ray studies is shown in Fig. II. 1. The Ca atoms lie in the invariant positions (0,0,0), with point symmetry $\overline{3}m$, and the oxygen and hydrogen atoms lie in the special positions $\pm(1/3,2/3,z_0)$ and $\pm(1/3,2/3,z_h)$ respectively, both with point symmetry $\overline{3}m$. The unit cell contains one molecule. The hydrogen positions, which could not be determined from the early X-ray work, were first postulated by Bernal and Megaw³. These positions have since been confirmed by X-ray diffraction^{7,8}, neutron diffraction¹⁰, and nuclear magnetic resonance¹¹.

The $Ca(OH)_2$ structure consists of layers of two sheets of hydroxyls parallel to the basal plane (0001), with a sheet of Ca atoms between them. Each Ca atom lies between six OH groups forming an octahedron which is slightly compressed along [0001]. The layers are held together by weak Van der Waals' forces between the opposed hydroxyl sheets. Neutron diffraction studies reveal that the hydrogen motion is confined mainly to the hydrogen layers normal to the c axis. Consequently, there are no hydrogen bonds. This is supported by the observed softness (2 on Moh's scale) and perfect cleavage in the basal plane. The H motion is apparently very anharmonic^{15,10}.

Structural data are collected in Table II.I. Fig. II. 2 shows in perspective the nearest- and second nearest-neighbour environment of



Fig.II.1. Structure of $Ca(OH)_2$ projected on (0001) plane. Some elements of the space group, D_{dd}^2 , are shown within the unit cell. These are: — - mirror plane, \triangle three-fold rotation axis. The calcium atoms are situated at inversion centers.



Fig.II.2. Nearest- and next nearest-neighbor environment of calcium atom in Ca(OH)₂. Unit cell dimensions at $T=293^{\circ}$ K. are shown. The angle $\delta = 61^{\circ}$ at $T=293^{\circ}$ K. ($\delta = 54.7^{\circ}$ for a cubic octahedron).

TABLE 1

Structural and other Data for $Ca(OH)_2$ Crystal

source	data						
Ref. 2	Linear thermal expansion $\alpha = (0.98 \pm 0.08) \times 10^{-5} \text{deg}^{-1}$	coefficients a_=(3.34 ⁺ 0.20);	10 ⁻⁵ deg ⁻¹				
Ref. 10		т=133 ⁰ К	т=293 ⁰ К				
Ker. 10	unit cell parameter 'a' unit cell parameter 'c' Ca-O distance (R _O) structure parameter z _O	3.5862+0.0006 Å 4.8801 0.0017Å 2.366 0.001 Å 0.2346 0.0002	3.5918+0.0003Å 4.9063 0.0007Å 2.371 0.001 Å 0.2341 0.0003				
	Unit cell dimensions were taken from Ref. 4, converted from KX to Å units, and corrected for thermal expansion using the coefficients of Megaw (Ref. 2).						
This work.	Effective thermal expansion parameter z_0 , $\alpha_{Z_0} = -1.3$ assuming usual exponentia and Levy data (Ref.10) way Temperature dependence be calculated from the for cell geometry. $R_0 = (1/3 a^2 + z_0^2 c^2)^{1/2}$ Debye Temperature θ = temperature specific heat obtained from Ref. 12.	on coefficient for 75 x 10^{-5} deg ⁻¹ , was 1 temperature dep s used at 133°K as ce of R ₀ and γ (so 110wing formulae $\gamma = ARCTan(a/T)$ = 260°K was calcus formula c = 2.60°K	or the position as calculated bendence. Busing and 293 ^o K. see Fig. II.2) car derived from unit (32 ₀ c) lated from low 54 x 10 ⁻⁵ T ³				

the calcium atom.

II. LATTICE DYNAMICS

The two OH groups in the unit cell make possible a highly complex phonon spectrum. Infra-red (IR) and Raman spectroscopy¹³ reveal the presence of OH rotatory modes and Ca-OH vibratory modes in addition to the fundamental OH stretching mode. Combinations of these also exist. Inelastic neutron scattering (INS) in the powder¹⁴ and in the crystal¹⁵ has also yielded evidence of some of these modes in addition to an acoustical branch of normal mode. Mitra¹⁶ and Oehler and Gunthard¹⁷ have explained the combinations theoretically.

We present in this section a short summary of results from these optical investigations which will be pertinent to our problem. Specifically, we concentrate on those $\mathbf{k} = 0$ optical modes which we expect to be dominant in the Mn²⁺ vibronic coupling.

According to factor group analysis, the dispersion surface of $Ga(OH)_2$ with 5 atoms per unit cell, has 15 distinct branches, of which 10 correspond to optical lattice modes, 3 to acoustical modes, and 2 to $\zeta^{\tau \tau'}$ internal OH modes. The physical nature of the 12 optical modes is sketched in Fig. II.3. The first Brillouin zone of $Ga(OH)_2$ is shown in Fig. II. 4. The symmetry elements of each little group G_k being known, it is possible to calculate the characters of the translation group representations using formulae given by Mitra¹⁸ and Oehler and Gunthard²⁰. It is customary, however, to list the number of times each of the latter representations is contained in the little group representations, rather



Fig.II.3. T' optical modes of vibration of $Ca(OH)_2$ crystal for k=0 (from Ref.18).



Fig.II.4. The first Brillouin zone of the trigonal lattice $(r < s/\sqrt{2})$ (from Ref.19). The little group at $\Gamma(0,0,0)$ is D_{3d} , at $\Delta(k_x,0,0)$ is C_2 , at $\Xi(0,k_y,0)$ is C_s , and at $\Lambda(0,0,k_z)$ is C_{3v} .

than the explicit translation group character table. This gives directly the number of modes of each type. Results of a partial factor group analysis 17 for the little group D_{3d} is given in Table II. II.

We expect only those modes which involve the relative motion of $Mn^{2+}(Ca)$ and OH to be responsible for any Mn^{2+} vibronic coupling. The frequencies of these modes, the so called T'-modes, are listed in Table II. III¹⁷. This point will be discussed further in Chapter III.

TABLE	II.II
-------	-------

Types of Modes (symmetry coordinates) for Ca(OH) Crystal

little group	wave vector k	modes						
G k	of phonon	irred. rep.	type					
		61 6 k	n i	Т	T'	R'	n' 1	
D _{3d}	$k_{x} = k_{y} = k_{z} = 0$	A _{1g}	2	0	1	0	1	
		A _{lu}	Ó	0	0	0	0	
		A _{2g}	0	0	0	0	0	
		A 2u	3	1	1	0	1	
		Eg	2	0	1	1	0	
		Eu	3	1	1	1	0	

TABLE II.III

د.

Fundamental	Т'	(k≈0)	Mode	Frequencies
-------------	----	-------	------	-------------

mode type	$\mathbf{\hat{k}}=0$ symmetry property	mode frequency cm ⁻¹
T'	A ₁	282
T'	E	247
T'	A ₂₁₁	321
T'	Eu	421

CHAPTER III

THEORY

I. CRYSTAL FIELD THEORY OF THE Mn²⁺ ION

The ground state of the free Mn^{2^+} ion is $3d^5 \, {}^6S_{5/2}$ in Russell-Saunders coupling. The ground state of the ion in a weak crystalline electric field of cubic symmetry is $(t_{2g})^3(e_g)^2 \, {}^6A_{1g}$. Neglecting spin, group theory predicts no splitting of this state in a crystal field since the state is orbitally non-degenerate. Physically, we would expect the 6-fold spin degeneracy to remain, since an electric field cannot affect spins directly. Since we observe EPR only in this ground state, the lack of orbital angular momentum would normally lead us to anticipate no crystalline field interactions. Contrary to expectations, however, the ground state of Mn^{2^+} in a cubic environment, such as that of MgO, splits into one Kramers doublet and one quartet²¹. The ground state of Mn^{2^+} in lower symmetries splits into three Kramers doublets. In addition, there is almost always a large hyperfine structure.

Bethe, in 1929, had showed that by means of the double group the state ${}^{6}A_{1g}$ in cubic symmetry decomposes to Γ_7 (2-fold) and Γ_8 (4-fold) levels²². In lower symmetries the Γ_8 level splits further into two 2-fold levels. In spite of the fact that crystalline field splittings of the S-state were rigorously predictable by group theory, physical mechanisms responsible for the splittings remained elusive. As we shall see in part IIIA of this chapter, the total crystalline field splitting is, in general, the result of the combined actions of the crystalline potential

 (V_c) , spin-orbit coupling (W_{LS}) , and spin-spin coupling (W_{SS}) . The splitting is also affected by lattice vibrations which modulate the crystalline electric field. In part IIIB we review S-state hyperfine interactions, and in part IIIC, the theory of the superhyperfine interaction pertinent to this work.

II. THE HAMILTONIAN FORMALISM AS APPLIED TO Mn²⁺ IN Ca(OH),

We anticipate that the Mn^{2^+} ion substitutionally replaces the Ca^{2^+} ion in the $Ca(OH)_2$ lattice. The point symmetry of the Mn^{2^+} ion should therefore be D_{3d} . A spin-Hamiltonian which is invariant in this symmetry and which accounts for all sources of Mn^{2^+} spin potential energy, is²³

$$H(s) = \beta \overline{H} \cdot \overline{g} \cdot \overline{s} + \overline{1} \cdot \overline{A} \cdot \overline{s} + \frac{1}{3} b_2^0 o_2^0 + \frac{1}{60} b_4^0 o_4^0 + \frac{3}{40} o_4^3 o_4^3$$
(III.1)

The terms are written in decreasing order of magnitude. The superhyperfine term has been omitted. The first term in Eq. (III.1) is the Zeeman term; the second is the hyperfine term; the remainder are the so-called crystal field terms. The 0^m_n are functions of spin operators and have been listed by Orbach²⁴. We assume that the g-tensor, the hyper-fine tensor, and the crystal field tensors, all have the same set of principal axes. We further assume that the g-tensor and the A-tensor have axial symmetry with principal values, g_{11} , g_{1} , and A and B respectively.

In order to diagonalize the Zeeman term, and thus simplify the calculation of matrix elements of Eq. (III.I), we transform to a coordinate system whose z axis, oriented at (β,α) relative to the crystal field axes, is the quantization axis of the electron spin. Transformation

formulae for 0_2^0 , 0_4^0 , and 0_4^3 have been given by Vinokurov et. al.²³ and by Thyer et. al.²⁵. The eigenvalues of Eq. (III.1) to second order perturbation in crystal field terms and hyperfine terms (including cross terms) are

$$E_{Mm} = g\beta HM + d_{2}^{0} + d_{4}^{0} + KM_{m} + \frac{1}{g\beta H_{0}} \left\{ \sum_{m=1}^{n} \frac{\left| d_{n}^{m} \right|^{2} - \left| d_{n}^{m} \right|^{2}}{m} + R + R_{m} \left(\frac{B^{2} - A^{2}}{2K} \right) \sin 2\beta + \frac{M[I(I+1) - m^{2}]}{4} \left(\frac{A^{2}B^{2}}{k^{2}} + B^{2} \right) + \frac{Mm^{2}}{2} \left(\frac{B^{2} - A^{2}}{2K} \right)^{2} \sin^{2} 2\beta - \frac{AB^{2}m}{2k} \left[s(s+1) - M^{2} \right] \right\},$$

where
$$d_n^{m-1} = \gamma a_n^{m-1} < M[0_n^m] M_{m}^{1}$$
 for $n = 2$, $m = 0, 1, 2; \gamma = 1/3$
 $n = 4, m = 0, 1, 2, 3; \gamma = 1/60$,
 $R = d_2^{1-}(d_4^{1-})^* - d_2^{1+}(d_4^{1+})^* + d_4^{1-}(d_2^{1-})^* - d_4^{1+}(d_2^{1+})^*$
 $+ 1/2 \ d_2^{2-}(d_4^{2-})^* - 1/2 \ d_2^{2+}(d_4^{2+})^* + 1/2 \ d_4^{2-}(d_2^{2-})^*$
 $- 1/2 \ (d_2^{2+})^* \ d_4^{2+}$,
 $R_x = d_2^{1-}d_x^+ + d_4^{1-}d_x^+ + d_x^-(d_2^{1-})^* + d_x^-(d_4^{1-})^* - d_2^{1+}d_x^- - d_x^{1+}d_x^-$
 $- d_x^+(d_2^{1+})^* - d_x^+(d_4^{1+})^*$, (III.2)

$$\mathbf{d}_{\mathbf{x}}^{+} = \langle \mathbf{M} | \mathbf{S}_{\mathbf{x}} | \mathbf{M}^{+} \mathbf{1} \rangle.$$

The $a_n^{m_{\pm}^{+}}$ have been listed by Vinokurov et al.²³. The elements $\langle M | 0_n^m | M_{\pm}^{+} | >$ have been extensively tabulated by Buckmaster²⁶ and Hutchings²⁷. From Eq.(III.2) we obtain the values of magnetic field at which resonances occur for constant frequency.

$$\begin{split} H\left(\frac{\pm}{2}, \mathbf{m} \rightleftharpoons \pm \frac{3}{2}, \mathbf{m}\right) &= H_{0} \ddagger 4a_{0}^{0} \ddagger 4a_{0}^{0} \ddagger 4a_{0}^{0} - \frac{1}{H_{0}} \left\{\frac{8}{9}|a_{2}^{1}|^{2} - \frac{4}{9}|a_{2}^{2}|^{2}\right\} \\ &- \frac{1}{40}|a_{1}^{1}|^{2} + \frac{1}{20}|a_{4}^{2}|^{2} + \frac{1}{120}|a_{4}^{3}|^{2} + \frac{1}{2}a_{2}^{1}(a_{4}^{1} + a_{4}^{1}) \\ &+ \frac{1}{2}a_{2}^{2}(a_{4}^{2} + a_{4}^{2})^{\pm}[2a_{2}^{1} + (a_{4}^{1} + a_{4}^{1})]\mathbf{m}\left(\frac{B^{2} - A^{2}}{2k}\right)\sin 2\beta \\ &+ \frac{1}{4}\left(\frac{35}{4} - \mathbf{m}^{2}\right)\left(\frac{A^{2}B^{2}}{k^{2}} + B^{2}\right)+\frac{1}{2}\mathbf{m}^{2}\left(\frac{B^{2} - A^{2}}{2k}\right)^{2}\sin^{2}2\beta\pm 2\mathbf{m}\frac{AB^{2}}{k}\right\} - \mathbf{m}K. \\ H\left(\frac{1}{2}, \mathbf{m} \rightleftharpoons \pm \frac{1}{2}, \mathbf{m}\right) &= H_{0} \ddagger 2a_{2}^{0} \ddagger 5a_{4}^{0} - \frac{1}{H_{0}}\left\{-\frac{1}{9}|a_{2}^{1}|^{2} + \frac{5}{9}|a_{2}^{2}|^{2} \\ &+ \frac{1}{5}|a_{4}^{1}|^{2} + \frac{9}{80}|a_{4}^{2}|^{2} - \frac{1}{120}|a_{4}^{3}|^{2} + \frac{1}{20}|a_{4}^{4}|^{2} - \frac{1}{2}a_{2}^{1}(a_{4}^{1} + a_{4}^{1}) \\ &+ \frac{1}{4}a_{2}^{2}(a_{4}^{2} + a_{4}^{2}) \pm \left[a_{2}^{1} - \frac{5}{4}\left(a_{4}^{1} + a_{4}^{1+}\right)\right]\mathbf{m}\left(\frac{B^{2} - A^{2}}{2k}\right)\sin 2\beta\pm \frac{1}{4}\left(\frac{35}{4} - \mathbf{m}^{2}\right) \\ &\left(\frac{A^{2}B^{2}}{k^{2}} + B^{2}\right) + \frac{1}{2}\mathbf{m}^{2}\left(\frac{B^{2} - A^{2}}{2k}\right)^{2}\sin^{2}2\beta\pm \mathbf{m}^{2}\mathbf$$

Eq. (III.1) is identical to that used by Bleaney and Trenam (1954)²⁸ for $b_2^0 = D$, $b_4^0 = \frac{1}{3}(F-a)$, $b_4^3 = \frac{20\sqrt{3}}{3}a$. (III.4)

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III. CALCULATION OF SPIN-HAMILTONIAN PARAMETERS

A. Crystal Field Parameters and g-values

(i) Static Splittings: The Point-Multipole (PM) Model

Calculations of the crystal field parameters and the g-values pertaining to d⁵ ions in ⁶S ground states involve the Zeeman interaction $(W_{\rm H})$, $V_{\rm c}$, $W_{\rm LS}$, and $W_{\rm SS}$. Various combinations of these interactions have been used in perturbation calculations up to as high as the sixth-order. The reasons such high order perturbations are needed are four-fold. (1) Matrix elements of any even-order $V_{\rm c}$ and matrix elements of $W_{\rm LS}$ within any electrostatic term individually vanish²⁹ so that $V_{\rm c}$ and $W_{\rm LS}$ cannot cause splitting in the first order. (2) $V_{\rm c}$ cannot, by itself, cause ground state splitting in any order of perturbation since the ⁶S ground state is the only state with spin multiplicity 6, and in the non-relativistic approximation, matrix elements of $V_{\rm c}$ vanish between terms with different spin multiplicities. (3) $W_{\rm LS}$ acting alone only shifts the ground state since it couples the ground state to only one other state, the ⁴P ³⁰. (4) $W_{\rm SS}$ acting alone only shifts the ground state state is couples the ground state to only one other state, the 4D ³¹.

If configurations other than d^5 are included, perturbations of lower order are possible. Pryce³² was the first to study the influence of V_c and W_{SS} via the ⁶D term of the higher-lying configuration $3d^44s$. In recent reviews, Sharma, Das, and Orbach^{33,34} (hereafter described as SDO) have extended the number of possible D-mechanisms (SDO used the 'D' notation, as used by Bleaney and Trenam²⁸, instead of the b⁰₂ notation

used in this work (see Eq. III.4). They have described these mechanisms collectively as point multipole (PM) mechanisms since they are based on the assumption that the lattice is wholly ionic. Since $Ca(OH)_2$ is highly ionic we shall not attempt to apply mechanisms based on covalency³⁸.

The SDO D-mechanisms, D_{BO} , D_{SS} , D_{ODS} , and D_{WC}^{33} are listed in Table III.I. The values of these quantities at room temperature (RT) were calculated using the formulae given by SDO. Excepting D_{BO} , these formulae contain B_2^0 , the coefficient of the axial potential, so that the application of these formulae to this problem involved the calculation of B_2^0 for the $[Ca(OH)_6]^{4-}$ complex. The point-charge lattice summation method was employed to compute B_2^0 since the structure of this complex is simple.

The result is:

$$D = D_{SS} + D_{ODS} + D_{WC}$$

= -0.07305B⁰₂ + 2.1044(B⁰₂)² - 1.7417(B⁰₂)². (III.5)

 B_2^0 for D_{3d} symmetry is:

$$B_{2}^{0} = \frac{-6e(3\cos^{2}\gamma - 1)}{B_{2}^{3}}$$
(III.6)

where R is the anion-cation distance, in units of the Bohr radius, and e is the electronic charge on the cation. γ is the angle between the c axis and the anion-cation bond direction.

Included in Table III.I is D_{VH} , the value of D calculated by van Heuvelen³⁵. By using relativistic Hartree-Fock wavefunctions for the basis states it can be shown that matrix elements of V_c no longer vanish between terms with different spin multiplicities. Although it has the

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TABLE III.I

6S-state Splitting Mechanisms Within The PM Model

rameter	General mechanism type Predicted inverse power of R ₀
6s-4 _T	$g^{(4_{\rm F},4_{\rm F},4_{\rm G})-4_{\rm T}} r_{\rm H}^{(4_{\rm F},4_{\rm F},4_{\rm G})-6_{\rm S}} s^{(4_{\rm F},4_{\rm F},4_{\rm G})-5_{\rm S}}$
d→ s,	d→d and d→g admixtures by ^W _{SS}
3d ⁵⁶ s- W	$(\underset{L_{S}}{\text{config}})^{4} r_{1g} [({}^{4} p_{j}^{4} r_{j}^{4} G_{V_{Q}}^{4} ({}^{4} r_{j}^{4} G_{j}^{4} P_{j}) \overline{v_{Q}}^{4} G_{j}^{4} P_{j}^{4} $
(A11 3	(d ⁵)(⁶ s- ⁴ P- ⁴ P- ⁴ P- ⁶ S) M_S v_2^0 v_2^0 M_S
(Relat	ivistic within $3d^5$) $(6s^4P6s)$ 3 $W_Ls^{V_0}_2$
12 dif 4th-,5	ferent mechanisms involving W _{LS} ,V _C ,W _{SS} , to th-,and 6th-orders, including spin doublets
(All 3	d^{5} b_{s}^{4} b_{w}^{-4} b_{s}^{-6} s_{s}^{-4} b_{w}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6} s_{s}^{-6}

T yalue 10 ⁴ cm ⁻ 1	~ +15	~ +10	~ +10	() spin- these , 8W,
Predicted R inverse x power of R ₀	even or odd ~ 10	-	-	field parameter Dq s. The values of quoted for D _{VH} , a _W
General Mechanism type	$^{W}\mathrm{_{LS}}$ and $^{W}\mathrm{_{SS}}$ to 6th-order in strong field approx.	Diag. of 3d ⁵ secular det.	Diagonalization neglecting W _{SS}	lculations are physical quantities, such as the crystal , Racah Parameters B and C, etc., assumed by the author umed vary slightly from reference to reference. Values directly from the references.
Parameter	ap	a G	ца В	t in these ca ng constant Ç hich were ass A were taken
Source	Ref. 40	Ref. 41	Ref. 42	<pre>* Implici * Implici orbit couplii quantities w ap, a_G, and</pre>

 $^{\mathrm{D}}_{\mathrm{BO}}$ was computed from the formula given by SDO using the point-charge lattice summation method. **

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same sign and comparable magnitude as our experimental D, we shall ignore $D_{\rm VH}$ in our further discussion.

Since we anticipate b_4^0 to be determined by 'a' to a good approximation, the table also includes contributions to 'a' coming from several possible mechanisms. It is impossible to calculate accurately the values of a_p , a_G , or a_L for the Ca(OH)₂:Mn²⁺ system since the separations in energy between the electrostatic terms are unknown.

(ii) Dynamic Splittings: Effects of Lattice Vibrations

(phonons)

We have seen in the previous section of this chapter that the crystal field parameters are functions of crystalline potentials of various symmetries. Since these potentials are linear combinations of spherical harmonics they involve the position coordinates of the source charges to various powers. When these coordinates are thermally modulated, the observed value of a parameter, G, is a time average, $\langle G \rangle_t$, which depends not only on the average value of the source coordinates (a change in this average value constitutes thermal expansion) but also on the detailed character of the vibrations. For example, for the simple case of harmonic modulation of the anion-cation distance, $d(t) = d_0(1+\varepsilon \cos \omega t)$, the time average value of d^n is $\langle d^n \rangle = d^n [1+n(n+1) \epsilon^2/4]$, $n \gtrless 0$. This means that a purely harmonic vibration of this type (A_{lg}) , while not affecting thermal expansion at all, does change the average value of the crystalline potential and thus also $\langle G \rangle_t^{43}$. Since the paramagnetic ion is subject to lattice vibrations the Hamiltonian (such as Eq. (III.1)) must be considered a 'dynamical' one which reflects the average site symmetry. Each

coefficient of this dynamical Hamiltonian is the average of a distribution which has been motionally narrowed and is not, in general, equal to the corresponding parameter in the static Hamiltonian³⁷. This notion of averaging a motionally narrowed distribution has been employed in describing the effect of lattice vibrations on optical spectra and on the chemical shift. The result is that the position of the spectral line depends explicitly on the temperature through its dependence on the meansquare amplitude of the lattice vibrations³⁷.

Assuming that the substitution of an Mn^{2^+} ion for a lattice cation does not disturb the normal modes or set up local modes, it can be shown that the functional form of the temperature dependence of some EPR Hamiltonian parameters is quite similar to that for, say, optical linewidths⁴⁴, and the chemical shift. There are two approaches. The most fundamental approach would be to calculate the normal modes of the $[Ca(OH)_6]^{4^-}$ system in a manner similar to Van Vleck's calculations in the alums when treating the Jahn-Teller effect⁴⁵ and spin-lattice relaxation⁴⁶. Two important points in this calculation for the alums were the following (1) Paramagnetic ions were subject to crystalline fields of predominantly cubic symmetry but with a small trigonal component. (2) The symmetry of the normal modes was O_h . Although (1) holds true for the $Ca(OH)_2:Mn^{2^+}$ system, (2) does not, since the normal modes in this crystal for k = 0 possess D_{3d} symmetry. (See Table II.II)

The 'vibronic' interaction, i.e., the interaction between the lattice modes and orbits of the Mn^{2+} electrons is given by the so-called orbit-lattice interaction

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$$W_{OL} = \sum_{i,j} V_{j}(\Gamma_{ig}) \varepsilon(\Gamma_{ig})$$
(III.7)

The $\epsilon(\Gamma_{ig}j)$ are the normal coordinates (displacement operators) of the $[Ca(OH)_{\vec{b}}]^{4-}$ system which are, in turn, those linear cominbations of components of the strain tensor which transform as the jth subvector of the representation Γ_{ig} . If Eq. (III.7) is assumed to operate only within d^5 , even vibrations need only be considered. We shall assume the Γ_{ig} to be those characterizing T' modes only. The Γ_{ig} are, of course, those T' representations which are permitted under one of the little groups of the space group D_{3d}^3 . The $V_j(\Gamma_{ig})$ are linear combinations of spherical harmonics (electronic operators) which transform as the Γ_{ig} . The components of the strain tensor are the normal coordinates q_{1m} associated with the lattice heat waves. These quantities introduce the temperature dependence, ie.

$$\varepsilon(\Gamma_{ig},j) = \sum_{\substack{m=1 \\ m=1}}^{3} \sum_{\substack{m=1 \\ m=1}}^{N} a_{j} q_{m}$$
(III.8)

where m denotes the phonon polarization, and N the total number of atoms in the crystal. The phonon operator is defined in the particle number representation by

$$< n_{j_{m}} |q_{j_{m}}| n_{j_{m}} + 1 > = < n_{j_{m}} + 1 |q_{j_{m}}| n_{j_{m}} > = [h(n_{j_{m}} + 1)/2\pi M_{v_{j_{m}}}]^{1/2}$$
(III.9)

where $\langle n_{\ell_m} \rangle = (1 - e^{-h v_{\ell_m}/kT})^{-1}$ and M is the mass of the crystal.

We terminate the discussion of the first approach by outlining the required calculations. We shall not attempt such calculations. To first order W_{OL} simply shifts the energy of the S ground state. Therefore,

one would naturally attempt to use W_{OL} as yet another perturbation term in a comprehensive treatment like that of Sharma, Das, and Orbach^{33,34} (see Table III.I). It can be seen from this table that W_{OL} , used in combination with W_{LS} and V_c , will affect D_{BO} and D_{WC} via the $\vec{k} = 0$ T' modes transforming as A_{1g} and E_g and will affect D_{ODS} via the $\vec{k} = 0$ T' modes transforming as A_{2u} and E_u . It is unlikely that a Waller-like mechanism will affect D_{SS} to any great extent since its effect in spinlattice relaxation is very small⁴⁶. Calculations such as these have not yet appeared in the literature. Their importance should be recognized mostly in non cubic crystals since experimentally it is known that the temperature dependence of 'a' is determined mainly by thermal expansion³⁷. We shall discuss the use of W_{OL} in more detail in the next section devoted to hyperfine structure.

The second approach, and the one we shall use for describing the temperature dependence of the crystal field parameters in this experiment, has been discussed by Bloembergen³⁷. We expand the parameter of interest, $G(\vec{r}_i - \vec{r}_0)$, in terms of the deviations $\vec{\delta}_i = \vec{\delta}r_i - \vec{\delta}r_0$ of the magnetic ion's Z effective neighbours located at $\vec{r}_i(i=1,2,\ldots Z)$ from their equilibrium positions $\vec{\Delta}_i = \vec{r}_i - \vec{r}_0$ relative to the ion at \vec{r}_0 . We assume that G is a scalar. To second order in the local coordinate deviations $\vec{\delta}_i$

$$G(\vec{\Delta}_{i}) = G_{0} + \sum_{i=1}^{Z} \left(\frac{\partial G}{\partial \vec{\Delta}_{i}}\right)_{0} \cdot \vec{\delta}_{i} + \frac{1}{2} \sum_{i,j=1}^{Z} \vec{\delta}_{i} \cdot \left(\frac{\partial^{2} G}{\partial \vec{\Delta}_{i} \partial \vec{\Delta}_{j}}\right) \cdot \vec{\delta}_{j} \quad (\text{III.10})$$

Lattice vibrations are introduced by expanding the $\vec{\delta}_i$ in terms of the normal coordinates $\vec{\xi}_i$

$$\vec{\delta}_{i} = \sum_{l} C_{il} \vec{\xi}_{l}$$
(III.11)

$$\vec{\xi}_{j} = \hat{\ell} \xi_{j0} \cos[\omega_{j} t - \vec{k}_{j} \cdot \vec{r} + \alpha_{j}(t)] \qquad (III.12)$$

Eq.(III.12) defined a travelling wave of amplitude ξ_{i0} , angular frequency ω_i , wave vector \vec{k}_i , time dependent phase α_i (due to finite phonon lifetimes) and unit polarization vector \hat{l} . Since the $\vec{\delta}_i$ are local coordinate changes the coefficients C_{il} weigh optical modes (specifically T' modes) most heavily. Substituting Eq. (III.11) and Eq. (III.12) into Eq. (III.10) and averaging over times long compared to phonon lifetimes (thus allowing the phases α_i to fluctuate randomly) we obtain:

$$c_{G_{t}} = c_{0} + \frac{1}{4} \sum_{i,j=1}^{Z} \sum_{i} \hat{l} \cdot \left(\frac{\partial^{2} c_{j}}{\partial \Delta_{i} \partial \Delta_{j}}\right) \cdot \hat{l} c_{il} c_{jl} \xi_{l0}^{2} \cdot (111.13)$$

This may be written in the simpler form

where

$$\langle G \rangle_{t} = G_{0} + \frac{1}{4} \int_{\ell}^{\Sigma} \left(\frac{\partial^{2} G}{\partial \xi_{\ell}^{2}} \right)_{0} \xi_{\ell}^{2}$$
(III.14)

Equating the energy of a lattice vibration to that of a quantum-mechanical harmonic oscillator leads to

so that Eq. (III.14) becomes

$$\langle G \rangle_{t} = G_{0} + \frac{\hbar}{4} \sum_{\ell} \left(\frac{\partial^{2}G}{\partial \xi_{\ell}^{2}} \right)_{0} \frac{1}{\omega_{\ell}} \text{ COTH } \left(\frac{\hbar\omega_{\ell}}{2kT} \right).$$
 (III.16)

We let the ℓ run over the T' optical modes. Using the frequencies given in Table II.III, and noting the degeneracies of the modes, Eq. (III.16) becomes, in the case when G = D,

$$D_{t} = D_{0}^{+4.97} \times 10^{-42} K_{1} \operatorname{coth}(\frac{196}{T}) + 11.3 \times 10^{-42} K_{2}$$
$$\times \operatorname{coth}(\frac{172}{T}), \qquad (111.17)$$

neglecting W_{OL} perturbations via configurational interactions. The K_{l} 's are the quantities $(\frac{\partial^2 D}{\partial \xi_{l}^2})_{0}$. I runs over the $A_{lg}(T')$ and $E_{g}(T')$ modes respectively. Thus by fitting Eq.(III.17) to the experimental curve we shall be able to determine D_{0} (the true 'static' D-value, the value of D in the absence of even zero-point vibrations) and the K_{l} 's.

B. Hyperfine Structure (HFS) Parameters A and B

(i) Static Splittings: Core Polarization

The HFS coupling constant A is a tensor which is the sum of the isotropic Fermi contact tensor and the traceless dipolar tensor. The large HFS which is usually observed for Mn^{2+} has been explained as follows. Originally, Abragam and Pryce⁴⁷ proposed a second order electrostatic interaction between 3d electrons and s-electrons of higher-lying configurations. Detailed work, however, has revealed that this mechanism gives HFS an order of magnitude too small. Heine⁴⁸ proposed that it was possible for the 2s and 3s core electrons to become slightly unpaired because of their exchange interaction with the 3d electrons. When the Mn^{2+} ion is in the ⁶S ground state, the spins of the five 3d electrons are oriented parallel(**1111**). Only those s-electrons which have like spin orientations

are affected by exchange. Consequently, the $2s(\uparrow)$ and $3s(\uparrow)$ electrons are attracted outward toward the 3d electrons by exchange forces leaving the $2s(\downarrow)$ and $3s(\downarrow)$ electrons to take part in the contact interaction. This effect is called core polarization⁴⁹. Calculations of A based on core polarization will not be carried out. We shall attempt to explain only A as a function of T.

(ii) Dynamic Splittings: The Orbit-Lattice Interaction

Simanek and Orbach⁵⁰ have satisfactorily explained the temperature dependence of A of Mn^{2+} in cubic crystals such as MgO. They proposed that, in addition to core polarization which is by far the dominant mechanism and which is temperature independent, W_{OL} admixes excited configurations of the form $3d^4ns$ into the ground $3d^5$ configuration. In first order, W_{OL} couples admixed s-states which have spins parallel to the spin of the 3d ground state so that the core polarization is slightly reduced. Important points in the Simanek-Orbach theory are:

(1) Since W_{OL} is assumed to operate between configurations involving d and s electrons, only j = 2 terms of Eq.(III.7) are important.

(2) W_{OI} is taken as a perturbation to first order only.

(3) The mean square strains $\langle e^2 \rangle$ are obtained from Eq. (III.8) assuming cubic symmetry.

We quote their result.

$$A(T) = A(0) \left[1 - CT^{4} \int_{0}^{\theta/T} \frac{x^{3} dx}{e^{x} - 1}\right]$$
(III.18)

where C is a constant and θ is the average Debye Temperature.

The HFS tensor of Mn^{2^+} in non-cubic crystals is usually slightly anisotropic due to the effect of the dipolar interaction. Since the dipolar tensor is traceless, the contribution of the isotropic Fermi contact interaction is $(A_{xx} + A_{y} + A_{z})/3$. The A_{jj} are the HFS coupling constants which are obtained experimentally for magnetic field parallel to the x,y and z direction. For crystals with axial symmetry, this becomes

$$\frac{A+2B}{3}$$

Thus Eq. (III.18) should explain the temperature dependence of this quantity for the $Ca(OH)_2:Mn^{2+}$ system.

C. Superhyperfine Structure (SHFS) Parameters

SHFS is accounted for by adding to the spin-Hamiltonian of Eq.(III.1) the additional term

$$H = \sum_{N} \vec{1}^{N} \cdot \vec{A}^{N} \cdot \vec{S}$$
(III.19)

where \hat{S} is the resultant Mn^{2^+} electron spin and \hat{I} is the proton spin. The summation is over the neighbouring protons, i.e., 6 in number if we consider only the nearest neighbours of the Mn^{2^+} (see Fig. II.2). The interaction is magnetic and proceeds via chemical bonding. The type of bonding determines the angular properties of Eq.(III.19). The proton HFS tensor \tilde{A}^N may be separated into contact and dipolar parts so that eigenvalues of Eq. (III.19) may be written

$$H = A_{s} \sum_{N} \overline{I}_{Z}^{N} + A_{p} \sum_{N} \overline{I}_{Z}^{N} (3 \cos^{2}\theta_{\zeta} - 1) p = \sigma, \pi \qquad (III.20)$$

 $\theta_{
m c}$ is the angle between the external magnetic field and the p-bond axis.

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For an arbitrary direction of magnetic field (\hat{H}) through the $[Ca(OH)_6]^{4-}$ complex, we should expect to see $3^3=27$ lines if $A_p \neq 0^{51}$.

For the special case of \hat{H} ||c axis, we should expect 7 equally spaced lines with intensity ratios 1:6:15:20:15:6:1. This should also hold for an arbitrary direction of \hat{H} if A = 0. We shall attempt only simple computations of A_s and A_p from the experimental data. No theoretical analysis will be given.

CHAPTER IV

INSTRUMENTATION AND CRYSTAL PREPARATION

The variable temperature work was done using a spectrometer of the straight-detection type operated in the x-band⁵².

A crystal rotating mechanism developed earlier in this laboratory was employed for anisotropy studies at RT^{53} . This mechanism enabled the crystal to be rotated about a horizontal axis. Any arbitrary orientation of the crystal relative to the external magnetic field could thus be attained by coupling this mode of rotation with that of the magnet about a vertical axis. A plastic cavity with the inner surface sputtered with gold was used in the LN_2 -RT work. The high temperature cavity (RT-800°K) was of stainless steel, the inner surface being gold electroplated onto copper. Both cavities, whose loaded Qs were about 6000 and 4000, respectively, were operated in the TE₀₁₁ mode. They are illustrated in Fig. IV. 1.

Copper-constant an and platinum + 10% Rhodium vs. platinum thermocouples were used for measuring temperatures. Reference temperatures were LN₂ and RT, respectively; the latter was monitored continually to ensure accuracy. Since no special attempts were made to calibrate the thermocouples, accuracy is expected to be of the order of $\pm 2^{\circ}$ K. Also, no attempt was made to control temperature. The LN₂ temperature cavity was heated to RT with 3-4 amps. of heater current at 6 volts delivered by a small regulated DC supply. The current was adjusted to an arbitrary value and measurements were made when the temperature was steady. At constant



Fig.IV.la. LN₂-RT variable temperature cavity.



Fig.IV.lb. RT-800°K. variable temperature cavity.

magnetic field modulation levels, temperature stability of about $\pm 1^{\circ}$ K was realized after one hr. The high temperature heater was supplied with about 7-8 amps DC at 12 volts maximum. Temperature stability of $\pm 2^{\circ}$ K. was obtained with the high temperature cavity within 1 1/2 hr. Forced air cooling of the high temperature cavity kept the change in the cavity resonant frequency to about 25MHz, over its range of operation.

The magnet used was a 15 in. Varian equipped with Fieldial regulation. All field measurements were made with the full 6 in. gap. The large gap was required in order to make room for a large glass dewar which contained the LN, coolant for the low temperature cavity. When the high temperature cavity was employed the large gap also made elaborate external cooling for protection of the polefaces unnecessary. Unfortunately, this arrangement introduced problems with inhomogeneity. To correct partially for inhomogeneity, two proton MR magnetometers were used to plot the magnetic field at the center of the magnet gap as a function of the magnetic field at the poleface center. The oscillator coil of one magnetometer was held temporarily at the center of the magnet gap (with the cavity removed) while the oscillator coil of the other was placed at the center of one of the polefaces. The outputs of both magnetometers were fed through an electronic switch and superimposed on an oscilloscope screen. The frequency measurements for a number of Fieldial settings throughout the range of interest are given in Appendix A. The data show that the frequency relation obtained, freq. at gap center (KHz) = 0.99266x freq. at poleface (KHz) + 0.0036, ensured an accuracy of at least +0.5 KHz. (0.1G) in extrapolating the proton MR frequency (field) at the

magnet gap center from the value measured at the center of the poleface. The sample was kept at least to within 1/4 in. of the magnet gap center at all times. As it was, inhomogeneity of the magnetic field over the sample volume (~60mm³) was estimated to be about 50-100 mG.

The magnetic field was modulated at 100 KHz. by means of small coils placed next to the cavities (see Fig. IV. 1). Very little modulation power was needed with the LN_2 temperature cavity. On the other hand, about 25 watts of power was required to penetrate the stainless steel of the high temperature cavity.

Single crystals of $Ca(OH)_2$ doped with Mn^{2^+} were obtained by the slow diffusion method⁵⁴. In addition to excluding CO_2 to ensure proper crystal growth, as much of the free oxygen as possible was removed to prevent the oxidation of Mn^{2^+} to Mn^{3^+} in solution. The steps involved in the preparation were as follows.

1) To remove 0_2 and $C0_2$ gases, distilled water was boiled in air and then cooled in flowing N_2 atmosphere inside a polyethylene 'glove-box'.

2) The reagents, $CaC1 \cdot 6H_{2}^{\circ}$ and NaOH, were dried in desiccators for 6-7 days.

3) The reagents, in weight ratios equal to the ratio of their molecular weights, with a pinch of dopant, $MnC1_2$ $^{4H}2^0$, were introduced into the diffusion box inside the 'glove-box'.

4) To react with any free oxygen remaining, a small amount of hydroxylammine hydrochloride was introduced into each compartment of the diffusion box. The lid was then sealed keeping N_2 atmosphere above the solution.

5) As a further precaution, the diffusion box was removed from the 'glove-box' and was placed in a sealed polyethylene bag which could be purged with N_2 at intervals.

A typical diffusion box is shown in Fig. IV. 2. The procedure above had to be repeated several times before crystals of a suitable size and dopant concentration were obtained. Well-formed, transparent, hexagonal, crystals of dimensions 4mm x 4mm x 3mm developed in about 8 weeks. Steps (1), (3), and (4) were repeated for separate solutions containing $CoCl_2$, $CuCl_2$, $NiCl_2$, GdO_3 , and EuO_2 (some of these were first dissolved in HCl) as dopants. The characteristic EPR spectra of each ion has since been seen in their respective crystals at LN₂ temperature.

Unfortunately, it proved impossible to obtain suitable $Ca(OD)_2:Mn^{2+}$ crystals. Step (1) above using heavy water proved impractical. Several attempts at introducing Mn^{2+} to the solution using $MnCl_2*4H_2O$ resulted in heavy oxidation of the ion to Mn^{3+} .



Fig.IV.2. Diffusion box used in growing Ca(OH)₂ crystals.

CHAPTER V

EXPERIMENTAL PROCEDURE AND RESULTS

I. ANALYSIS OF THE RT SPECTRUM

The first step in the EPR work was to identify the type of impurity in $Ca(OH)_2$ and to confirm the predictions of the crystal structure and space group. Therefore, the angular variation of the spectrum was examined by rotating the crystal in the cavity, using the crystal rotating mechanism described in Chapter IV. A stereographic projection was employed to map the polar and azimuthal angles which located the magnetic field (\hat{H}) relative to the crystallographic axes. Typical recordings of the spectrum for \overline{H} parallel and perpendicular to the crystallographic c axis, [0001], are shown in Fig. V.3. The spectrum consisted of six groups of five lines indicative of HFS splitting larger than fine structure (FS) splitting. The spectrum was characteristic of only one ${\rm Mn}^{2^+}$ ion per unit cell. Angular variations of the spectrum are given in Fig. V.1. The spectrum was isotropic for rotation of \hat{H} in the (0001) plane, and was symmetrical about the [0001] axis for rotation of \hat{H} in the (1010) plane. The spectrum exhibited no symmetry for rotation of $\hat{\vec{H}}$ in the (1210) plane, nor in any other arbitrary plane. This behaviour is consistent with Mn^{2+} in Ca lattice sites having D_{3d} symmetry. These observations led us to choose the c axis as the magnetic z axis.

The resonant fields measured by proton MR are given in Table V.I. These measurements were used in a computer program which yielded the Hamiltonian parameters corresponding to the mimimum RMS error-of-fit

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Fig.V.la. Angular variation of spectrum for rotation of H in (1010) plane. T=290 K. The crosses are experimental measurements which were obtained for the microwave frequency 9380 MHz. The data for H [0001] and H [1010] are listed in Table V.l. The solid curves are plots of the formulae given in eqs.(III.3) for the values of the spin-Hamiltonian parameters given in Table V.l.



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TABLE V.I

Direc- tion of Ĥ	TRANSITION*			MAGNETIC FIELDS IN GAUSS		
	m	М	M-1	ME AS URE D	CALCULATED	ERROR
[0001]		-3/2	-5/2	3062.6	3063.1	+ 0.5
		-1/2	-3/2	3104.6	3104.9	+ 0.3
	- 5/2	1/2	-1/2	3112.4	3112.0	- 0.4
		3/2	1/2	3118.9	3119.1	- 0.2
	-3/2	1/2	-1/2	3199.8	3199.2	- 0.6
		5/2	3/2	3245.5	3243•2	- 2.3
	-1/2	1/2	-1/2	3289.0	3288.8	- 0.2
		5/2	3/2	3328.4	3328.0	- 0.4
	1/2	1/2	-1/2	3380.8	3380.9	+ 0.1
		5/2	3/2	3415.6	3415.1	- 0,5
	<u></u>	-3/2	-5/2	3446.0	3445.9	- 0.1
	3/2	1/2	-1/2	3474.9	3475.3	+ 0.4
		5/2	3/2	3505.1	3504.8	- 0.3
		-3/2	-5/2	3549.1	3547.7	- 1.4
	5/2	1/2	-1/2	3571.4	3572.3	- 0.9
		5/2	3/2	3596.2	2596.8	+.0.6

Position of Resonant Fields at Room Temperature $(290^{\circ}K)$ and for Microwave Frequency 9380 MHz. 'Calculated'fields were computed for the Spin-Hamiltonian parameters $g_{11} = 2.0011\pm0.0005, g_{12} = 2.0010\pm0.0005, b_{12}^{\circ} = -6.7\pm0.5G, b_{12}^{\circ} = -2.48\pm0.25G, b_{13}^{\circ} = 0\pm100G, A=-92.05\pm0.15G, B=-90.28\pm0.15G.$

Direc- tion of Ĥ	TRANSITION*			MAGNETIC FIELDS IN GAUSS		
	m	М	M-1	ME AS URE D	C ALCUL ATE D	ERROR
[10]0]	-5/2	1/2	-1/2	3113.2	3113.3	+ 0.1
	-3/2	1/2	-1/2	3198.2	3198.6	+ 0.4
		-3/2	-5/2	3291.5	3293.7	+ 2.2
	-1/2	1/2	-1/2	3286.7	3286.4	- 0.3
		5/2	3/2	3280.5	3279.3	- 1.2
		-3/2	-5/2	3386.6	3388.9	+ 2.3
	1/2	1/2	-1/2	3376.7	3376.6	- 0.1
		3/2	1/2	3366.5	3364.1	- 2.4
	<u></u>	-1/2	-3/2	3483.5	3484.5	+ 1.0
	3/2	1/2	-1/2	3469.4	3469.4	0.0
	3	3/2	1/2	3455.7	3454.4	- 1.3
	<u></u>	-1/2	-3/2	3582•4	3582.3	- 0.1
	5/2	1/2	-1/2	3564.5	3564.7	+ 0.2
		3/2	1/2	3547.3	3547.1	- 0.2

* Quantum numbers were assigned to the Lines on the basis of the work of Pieczonka et al.⁵⁵ on Mn^{2+} impurity in natural Brucite, $Mg(OH)_2$, which has a structure isomorphous with $Ca(OH)_2$.



Fig.V.2. Typical recordings of the spectrum at T=96^OK. (a) Recording for H || [0001]. (b) Energy level diagram for H || [0001]. (c) Recording for H || [1010]. (d) Energy level diagram for H || [1010]. Hyperfine m quantum numbers for each type of Δ M_g=±1 transition read -5/2 to +5/2 from left to right. Eigenvalues were plotted by computer from eqs.(III.2).



Fig.V.3. Typical recordings of the spectrum at RT. (a) $\hat{H} \parallel [0001]$. (b) $\hat{H} \parallel [1010]$.



Fig.V.4. Typical recordings of the spectrum at 609[°]K. (a) HN[0001] . (b) HN[1010] . The sharp central line in (a) is the resonance of the free radical dpph.

between the measured resonant fields and the calculated resonant fields. These parameters, are also included in Table V.I. They give an RMS error of 0.78G for 27 lines. II. THE FS AND HFS SPECTRUM AS A FUNCTION OF TEMPERATURE

In Figs. V.2 and V.4 we present recordings of the spectrum at temperatures 96°K. and 609°K., respectively. The spectra shown here and the ones shown in Fig. V.3 indicate that, as the temperature of the crystal varies from LN_2 temperature, through RT, to 609° K, the FS components of each HFS line reverse their order. For example, the FS components of the m = -5/2 line, for Hic, crossed over each another at about RT, whereas those of the m = +5/2 line did not do so until about 609° K. The FS components of the other HFS lines crossed over one another at temperatures between these two. This behaviour indicates that b_2^0 reverses sign.

The stability of the bridge spectrometer and cavity system was such that over the temperature range LN_2 -800°K. the spectrum could be observed at all times on the oscilloscope. At no time did the spectrum undergo abrupt changes which would be indicative of discontinuous changes in a Hamiltonian parameter. The spectrum was visible and still reversible for temperatures of short duration up to about 825°K. (see Fig. V.5). At temperatures in this region and above, however, the spectrum gradually died out and did not return upon cooling. When the crystal cooled from these high temperatures, it was observed to be partially decomposed into a white powder, presumably CaO. No EPR lines characteristic of the CaO:Mn²⁺ system were observed in this white powder above RT.

In 5 experimental runs, the spectrometer-cavity system was stabilized at 30 different temperatures. At each temperature the resonant fields were measured by proton MR. The results were used to compute the

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Fig.V.5. Recording of the spectrum at $T=825^{\circ}K$.

Hamiltonian parameters by the method described in the previous section. The results for b_2^0 , b_4^0 , A and B are given in Figs. V.6, V.7, and V.8, respectively. The fit was so insensitive to variations of b_4^3 that no meaningful temperature dependence of this parameter was obtained. The g-value remained constant to within the experimental error, ± 0.0005 , over the whole temperature range.

A. Temperature Dependence of the Crystal Field Parameters

Fig. V.6 shows that at temperatures higher than about 150° K,, the experimental curve is essentially linear. At low temperatures the curve tapers off to approach, with zero slope, the value -16.25G at T=0°K. This value has been extrapolated assuming the crystal does not undergo a phase change below LN₂ temperature.

The curve labelled 'a' is a plot of the D function predicted by SDO, given by Eq. (III.5), corrected to units of gauss. Two important assumptions were made, (1) that the anion-cation distance is R_0 (the Ca-O distance), and (2) that the oxygen possesses one electronic charge. Temperature dependences of R_0 and γ were computed from the data in Table II.I.

The fitting of Eq. (III.17) to the experimental curve was simplified by first computing approximate values of D_0, K_1 and K_2 . A straight line visual fit to the high temperature data was drawn, the line labelled 'b' in Fig. V.6, and from it the intercept, -21.50G, and slope, $4.95 \times 10^{-2} G/K$, was obtained. Since the high temperature approximation of Eq. (III.17) is

$$D(T) = D_0 + \left(\frac{4.97 \times 10^{-42} K_1}{196} + \frac{11.3 \times 10^{-42} K_2}{172}\right) T,$$

we have the relations





Fig.V.7. b_4^0 vs. temperature. Possible errors in the data are $\pm 0.25G$. Curves 'a' and 'b' are plots of the functions $b_4^0=M_1R_0^{-5}$ and $b_4^0=M_2R_0^{-10}$ where $M_1=-4.12 \times 10^3$ and $M_2=-7.40 \times 10^6$. Units of M_1 and M_2 are gauss $\times a_0^5$ and gauss $\times a_0^{10}$ where a_0 is the Bohr radius. These functions represent typical mechanisms which have been proposed for the parameter a taken from Table III.1.

$$D_0 = -21.50G$$
,

and
$$\frac{4.97}{196} \times 10^{-42} K_1 + \frac{11.3}{172} \times 10^{-42} K_2 = 4.95 \times 10^{-2}$$

In addition, the T=O approximation of Eq. (III.17) gives

$$D_0 + 4.97 \times 10^{-42} K_1 + 11.3 \times 10^{-42} K_2 = -16.25$$

so that by solving these equations we obtained the values $D_0^{=-21.50\pm0.50G}$, $K_1^{=(5.38\pm0.5)\times10^{42}G/erg-sec^2}$ and $K_2^{=(2.83\pm0.5)\times10^{42}G/erg-sec^2}$. A plot of Eq.(III.17) using these values is shown as curve 'c' in Fig.V.6. An attempt was made to improve on these values by using them as zeroth order approximations in a best fit program. The results however, gave a curve which favoured the low temperature data while giving a poor fit to the high temperature data. The quantities D_0 , K_1 and K_2 quoted above give the best fit of Eq. (III.17) over the whole temperature range.

Since the temperature dependences of b_2^0 and of A were qualitatively similar, an expression for b_2^0 , of the same form as that for A, given by Eq. (III.18), was also fitted to the experimental data. Normalizing the two curves at 636°K. and using the values $b_2^0(0) = -16.25$ G, $\theta = 260^{\circ}$ K. we obtained the value 5.07 x 10^{-100} K⁻⁴ for the constant C in Eq.(III.18). These values give curve 'd' in Fig. V.6. A slightly better fit at low and high temperatures was obtained for the values $b_2^0(0)=-16.25$ G, $\theta=500^{\circ}$ K., and C=8.28x10⁻¹¹⁰K⁻⁴. This is curve 'e' in Fig. V.6.

The method of using a best fit program to derive the Hamiltonian parameters resulted in a non-uniform variation of b_4^0 values as seen in Fig. V.7. The RMS error of fit was far more sensitive to variations in

 b_2^0 , A and B than in b_4^0 . Thus it was impossible to define an accurate fit to the data. The curves labelled 'a' and 'b' represent the functions $b_4^0 = M_1 R_0^{-5}$ and $b_4^0 = M_2 R_0^{-10}$, respectively, where M_1 and M_2 are constants. These functions were normalized to the value of $b_4^0 = -2.3G$ at $T = 200^{\circ}K$. B. Temperature Dependence of the Hyperfine Parameters

The temperature dependences of A and B have been given in Fig.V.8. We have not attempted a theoretical analysis of A(T) and B(T) taken separately. Qualitatively, the curves were similar. The quantity $\eta = \left| \frac{(A^2 - B^2)^{1/2}}{A} \right|$, taken from the data of Fig. V.8, is shown as a function of temperature in Fig. V.9. A straight line visual fit to the data is indicated. The rate at which η decreased, given by the slope of this line, was of the order of 7.6 x 10^{-5} °K⁻¹. The temperature dependence of the quantity (A+2B)/3, obtained from the data of Fig. V.8, is given in Fig. V.10. Eq.(III.18) was fitted to the experimental curve by evaluating the integral numerically and normalizing the theoretical curve and experimental curve at 600°K. The best fit for a Debye Temperature of 260°K was obtained for A(0) = -92.05G and $C = 10.6 \times 10^{-12} \text{ oK}^{-4}$. A plot of Eq. (III.18) for these constants is the curve labelled 'a' in Fig. V.10. The best overall fit to the data, however, was obtained for A(0) = -92.00G, $\theta = 500^{\circ}K$., and $C = 1.78 \times 10^{-120} k^{-4}$. These values give the curve labelled 'b' in Fig. V.10.

III. Mn²⁺-H SHFS AND GENERAL DISCUSSION OF LINESHAPES

The Mn^{2^+} -H SHFS was studied in the (1010) plane at a temperature of 85°K and for HLc in the range 80-130°K.

Fig. V.11 shows SHFS splittings of the FS components of the

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Fig.V.8. A and B vs. temperature.



Fig.V.9. $\eta = \left| \frac{(A^2 - B^2)}{A} \right|^2$ vs. temperature. The data are listed in Appendix B.



Fig.V.10. (A+2B)/3 vs. temperature. Data are listed in Appendix B.



Fig.V.ll. FS components of the m=+5/2 hyperfine line. (a) $H \parallel [0001]$. (a) $H \parallel [1010]$.

m = +5/2 hyperfine line. The physical properties of the corresponding FS components of the other hyperfine lines were similar, as can be seen from Fig. V.2. For both orientations of \hat{H} , the FS components were displaced symmetrically about the $M_s: -\frac{1}{2} \leftrightarrow +\frac{1}{2}$ line. For \hat{H}_{LC} the intensity ratios were approximately 5:11:15:11:5 (taking peak-to-peak heights as proportional to the line intensities, which is true if the line widths are the same). For \hat{H}_{LC} the intensity ratios were approximately 5:8.7: 14:27:11 but the line widths were not equal. Neglecting SHFS the ratios should be 5:8:9:8:5. The broadest line at 85°K was of the order 6G wide. This was the $M_s: -5/2 \leftrightarrow -3/2$, m = 5/2 line with unresolved SHFS for \hat{H}_{LC} .

Fig V.11 shows that not all FS components were split equally. For \widehat{H}_{LC} correspondingly displaced FS lines were split equally. For \widehat{H}_{HC} , however, the M_s: $+3/2 \leftrightarrow 1/2$ line was split the most, followed by the lines M_s: $5/2 \leftrightarrow 3/2$, M_s: $1/2 \leftrightarrow -1/2$, and then the rest, which showed no resolved structure at all.

Fig. V.12 shows the $M_s: 1/2 \leftrightarrow 3/2$, m = -5/2 line for \hat{H} at low (~1mW) and high (~50mW) power levels. At low power a 5 line spectrum was fully visible. At high power there existed just a hint of a 7 line spectrum. The central component showed some saturation at high power (full power from an X13 Klystron) at 85° K. The approximate intensity ratios at low power were x:5:37:95:39:7:x (x was unknown at low power) as compared with the theoretical ratios 1:6:15:20:15:6:1. From chart recordings obtained at high power (Fig. V.12) the components were observed to be equally spaced by approximately 4G.



Fig.V.12. The M_s: 1/2 \leftrightarrow 3/2; m=-5/2 line at low and high power levels. (a) low power (~1 mW). (b) high power (~50 mW).

The angular variation of the relative SHFS splitting between the central component and the first components on either side of the central component is shown in Fig. V.13. Each point is the mean of about 12 separate measurements, the 'upper' and 'lower' splittings of the FS components of each of the 6 hyperfine lines. Data are missing between about 20° and 65° due to slight broadening and considerable overlapping of the lines. A dependence of this splitting on M_s is evident from the figure. There also appears to be a slight anisotropy. The mean value of the splitting for $\widehat{H} \parallel c$ indicates $A_s \sim 4.5G$. Finally, the temperature dependence of the relative SHFS splittings was linear, within the experimental error, in the range 80° K-130°K. Above 130°K SHFS splittings were obscured by slight broadening and overlapping of the lines.

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Fig.V.13. Angular variation of SHFS splitting relative to the central component. Rotation of H is in (1010) plane. Legend: 0 - 3/2 ↔ 5/2 FS component; ● - 1/2 ↔ 3/2 FS component; □ - -1/2 ↔ 1/2 FS component; ■ - -3/2 ↔ -1/2 FS component.

CHAPTER VI

DISCUSSION AND CONCLUSIONS

The temperature dependences of the spin-Hamiltonian parameters showed no discontinuities in the temperature range 85° K- 800° K. We can conclude, therefore, that the Ca(OH)₂ crystal remained stable in the trigonal phase in this range. The fact that the crystal began to decompose at temperatures of 800° K and above, is consistent with the fact that its dissociation pressure reaches 1 atm. at 785° K.⁵⁶ The Mn²⁺ probably oxidized to Mn³⁺ during this dehydration reaction since no EPR lines were observed in the powder product of the dehydration.

The $b_2^0(T)$ which was predicted from the theory of SDO (curve 'a' in Fig. V.6) coincided with the b_2^0 which was measured experimentally at 150°K. Also, the value of $b_2^0(0) = -14.25$ G, which was predicted by this theory, differed by only 2G from the value of $b_2^0(0) = -16.25$ G which was extropolated from the experimental data. We conclude, therefore, that the SDO theory gives a good fit to the experimental data at low temperatures. It is possible that a better fit could have been obtained had the OH group been treated as a dipole, instead of as a point charge located on the oxygen, and had a more accurate value of the anion-cation bond distance been determined. The value of B_2^0 in the point-charge computation could also be improved by summing eq. (III.6) over many neighbours. In any case, the theoretical function $b_2^0(T)$ should remain a linear function of temperature

because of the thermal expansion of R.

Proceeding on the assumption that the $b_{2}^{0}(T)$ predicted by SDO is the correct explanation of the experimental $b_2^0(T)$ at low temperatures, the experimental $b_2^0(T)$ should have followed the curve 'a' in Fig. V.6 at high temperatures, if the temperature dependence were determined only by the thermal expansion of R. The fact that the two curves differed markedly at high temperatures implies that thermal expansion is not of major importance in the experimental $b_2^0(T)$. Unfortunately we were not able, in a simple manner, to estimate the contribution of thermal expansion to the data. This is only possible when the spin-Hamiltonian parameters are studied as a function of both temperature and pressure.³⁷ The fact that the $b_2^0(T)$ predicted from the theory of Walsh, Jeener, and Bloembergen³⁷, given by Eq. (III.17), gave a good fit to the data shows that discrete lattice modes can account for the experimental $b_2^{(0)}(T)$. The assumption that the T' modes transforming as A and E in D symmetry are the modes most responsible for the tem-lg 3dperature dependence of $b_2^0(T)$ thus gives satisfactory results. The fact that a curve of the form of Eq. (III.18) gave fits to the data almost as good as the above means that it is impossible to state whether the Mn^{2^+} vibronic coupling is due exclusively to lattice modes of certain discrete frequencies, or to a continuous spectrum of modes describable by a Debye approximation. Both approaches appear to be equally valid.

We feel that we have been justified in applying to the data an expression for $b_2^0(T)$ of the same form as that for A(T) which was derived by Orbach and Simanek⁵⁰. The mechanisms which apply for $b_2^0(T)$ will, of course, be different from the Fermi contact interaction which applies for

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A. In any case, a function of the general form of Eq. (III.18) should result from computations involving W_{OL} since the form of Eq. (III.18) is the result of averaging the mean square strains.

Although the accuracy in the determination of b_4^0 was poor, the temperature dependence of b_4^0 tended to show a linear behaviour. Therefore, the effects of lattice vibrations on the experimental $b_4^0(T)$ were probably negligible. Walsh et. al.³⁷ reached the same conclusion for the parameter a. Although the experimental $b_4^0(T)$ tended to show a dependence, such as $b_4^0 = CR_0^{-n}$, where C is a constant, R_0 is the Ca-O bond distance and n is some number (see Table III.I), it proved impossible to determine the value of n unequivocally.

The theory of Simanek and Orbach⁵⁰ gave a good fit to the isotropic part of the axially-symmetric HFS tensor. As expected, the temperature dependence of this quantity was similar to the temperature dependence of the isotropic HFS tensor of Mn^{2+} in cubic crystals⁵⁰. The best fit to the data was obtained for an average Debye Temperature of 500° K instead of for the value of 260° K which was calculated from the low temperature specific heat data. We should expect an average Debye Temperature to be somewhat higher than a Debye Temperature which is derived from low temperature specific heat data since the Debye Temperature is somewhat temperature specific heat data since the Debye Temperature is somewhat temperature dependent. The constants of Eq. (III.18) which were determined in this work, namely, $\theta = 500^{\circ}$ K., $C = 1.78 \times 10^{-120}$ K⁻⁴, compare favourably with the values found for Mn^{2+} in CaO, namely, $\theta = 450^{\circ}$ K., and $C = 2.30 \times 10^{-120}$ K⁻⁴.⁵⁷ The fact that η tended to decrease with temperature implies that the axial distortion of the HFS tensor tended to decrease with temperature. There is no

60

explanation in the literature to account for this effect.

The SHFS tensor was largely isotropic. The isotropic Fermi contact part was of the order of 4.5G at 85° K. The anisotropic dipolar part was probably of the order of 0.1G or less. There appeared to be a slight dependence of this splitting on M_s. This could have been the result of a second-order effect.
REFERENCES

1.	Early references are quoted by R.W.G. Wyckoff, <u>The Structure of</u> <u>Crystals</u> (Reinhold Pub. Comp., N.Y. 1931).
2.	H.D. Megaw, Proc. Roy. Soc. (London), <u>A142</u> , 198 (1933).
3.	J.D. Bernal and H.D. Megaw, Proc. Roy. Soc. (London), <u>A151</u> , 384 (1935).
4.	Bunn, Clark, and Clifford, Proc. Roy. Soc. (London), <u>A151</u> , 141 (1935).
5.	H.E. Petch and H.D. Megaw, J. Opt. Soc. Am., <u>44</u> , 744 (1954).
6.	H.E. Petch, Phys. Rev., <u>99</u> , 1635 (1955).
7.	H.E. Petch, Can. J. Phys., <u>35</u> , 983 (1957).
8.	H.E. Petch, Acta, Cryst., <u>14</u> , 950 (1961).
9.	H.E. Swanson and E. Tatge, Nat. Bur. Stand. Cir. 539, <u>1</u> , 58 (1953).
10.	W.R. Busing and H.A. Levy, J. Chem. Phys., <u>26</u> , 563 (1957).
11.	D.M. Henderson and H.S. Gutowsky, Am. Mineral., <u>47</u> , 1231 (1962).
12.	W.E. Hatton, D.L. Hildenbrand, G.C. Sinke, and D.R. Stull, J. Am. Chem. Soc., <u>81</u> , 5028 (1959).
13.	R.A. Buchanan, H.H. Caspers, and J. Murphy, Applied Optics, <u>2</u> , 1147 (1963).
14.	G. Safford, V. Brajovic, and H. Boutin, J. Phys. Chem. Solids, <u>24</u> , 771 (1963).
15.	I. Pelah, K. Krebs, and Y. Imry, J. Chem. Phys., <u>43</u> , 1864 (1965).
16.	S.S. Mitra, J. Chem. Phys. <u>39</u> , 3031 (1963).
17.	0. Oehler, and H.H. Gunthard, J. Chem. Phys., <u>48</u> , 2036 (1968).
18.	S.S. Mitra, Solid State Physics, <u>13</u> , 1 (1962).
19.	G.F. Koster, Solid State Physics, <u>5</u> , 1 (1957).
20.	0. Oehler and H.H. Gunthard, J. Chem. Phys., <u>48</u> , 2032 (1968).

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21. W. Low, Phys. Rev., <u>105</u>, 793 (1957).

• •

22.	H.A. Bethe, Ann. Physik, <u>3</u> , 133 (1929); English translation, Consultants Bureau, New York, 1958.
23.	V.M. Vinokurov, M.M. Zaripov, and V.G. Stepanov, Soviet Phys Solid State, <u>6</u> , 866 (1964).
24.	R. Orbach, Proc. Roy. Soc. (London), <u>A264</u> , 458 (1961).
25.	J.R. Thyer, S.M. Quick, and F. Holuj, Can. J. Phys. <u>45</u> , 3597 (1967).
26.	H.A. Buckmaster, Can. J. Phys., <u>40</u> , 1670 (1962).
27.	M.T. Hutchings, Solid State Physics, <u>16</u> , 227 (1964).
28.	B. Bleaney and R.S. Trenam, Proc. Roy. Soc. (London), A223, 1 (1954).
29.	H. Watanabe, <u>Operator Methods in Ligand Field Theory</u> (Prentice-Hall, Inc., N.J. 1966).
30.	G. Racah, Phys. Rev., <u>63</u> , 367 (1943).
31.	R. Trees, Phys. Rev., <u>82</u> , 683 (1951).
32.	M.H.L. Pryce, Phys. Rev., <u>80</u> , 1107 (1950).
33.	R.R. Sharma, T.P. Das, and R. Orbach, Phys. Rev., <u>149</u> , 257 (1966).
34.	R.R. Sharma, T.P. Das, and R. Orbach, Phys. Rev., <u>171</u> , 378 (1968).
35.	A. van Heuvelen, J. Chem. Phys., <u>46</u> , 4903 (1967).
36.	W.M. Walsh Jr., Phys. Rev., <u>122</u> , 762 (1961).
37.	W.M. Walsh Jr., J. Jeener, and N. Bloembergen, Phys. Rev., <u>139</u> , A1338 (1965).
38.	R.R. Sharma, T.P. Das, and R. Orbach, Phys. Rev., <u>155</u> , 338 (1967).
39.	H. Watanabe, Prog. Theoret. Phys., <u>18</u> , 405 (1957).
40.	M.J.D. Powell, J.R. Gabriel, and D.F. Johnston, Phys. Rev. Lett., 5, 145 (1960).
4 1.	J.R. Gabriel, D.F. Johnston, and M.J.D. Powell, Proc. Roy. Soc. (London) <u>A264</u> , 503 (1961).
42.	W. Low and G. Rosengarten, J. Mol. Spectr. <u>12</u> , 319 (1964).

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- 43. W.M. Walsh Jr., Phys. Rev., <u>114</u>, 1473 (1959).
- 44. C.C. Klick and J.H. Schulman, J. Opt. Soc. Am., <u>42</u>, 910 (1952).
- 45. J.H. Van Vleck, J. Chem. Phys., 7, 72 (1939).
- 46. J.H. Van Vleck, Phys. Rev., 57, 426 (1940).
- 47. A. Abragam and M.H.L. Pryce, Proc. Roy. Soc., (London), <u>A205</u>, 135 (1951).
- 48. V. Heine, Phys. Rev., <u>107</u>, 1002 (1957).
- R.E. Watson and A.J. Freeman eds., <u>Hyperfine Interactions</u>, (Academic Press N.Y. 1967).
- 50. E. Simaneck and R. Orbach, Phys. Rev., <u>145</u>, 191 (1966).
- 51. M. Tinkham, Proc. Roy. Soc., (London), <u>A236</u>, 535 (1956).
- 52. S.M. Quick, M.Sc. Thesis, U. of Windsor (1966) (Unpublished).
- 53. F. Holuj, Can. J. Phys., <u>46</u>, 287 (1968).
- 54. F.W. Ashton and R. Wilson, Am. J. Sci., <u>13</u>, 209 (1927).
- 55. W.A. Pieczonka, H.E. Petch, and A.B. McLay, Can. J. Phys., <u>39</u>, 145 (1961).
- 56. P.E. Halstead and A.E. Moore, J. Chem. Soc., (London), 3873 (1957).
- 57. R. Calvo and R. Orbach, Phys. Rev., <u>164</u>, 284 (1967).

APPENDIX A

Proton MR Frequencies Obtained in Calibration of Magnetic Field

Across 6 in. Gap of Varian Magnet

a

eq. (KHz) gap center	Freq. (KHz) from formula	Freq. (KHz) at pole face	Freq. (KHz) at gap center	Freq. (KHz from formul
	L1440.5	13819.7	13721.9	13721.
	11465.6	. 14040.4	13941.2	13940.9
	11573.0	14250.0	14149.0	14149.0
	11789.0	14468.5	14366.0	14365.9
	12005.7	14683.9	14580.0	14579.7
	12222.5	14898.8	14793.1	14793.C
	12431.0	15114.1	15006.8	15006.8
	12651.8	15330.1	15221.4	15221.2
	12870.6	15546.7	15436.2	15436.2
	13080.6	15761.4	14649.4	15649.3
	13291.7	15979.1	15865.5	15865.4
	13510.3	16192.6	16077.8	16077.3
		16498.3	16381.1	16380.8

APPENDIX B

Best fit data of spin-Hamiltonian parameters b_{4}^{0} , b_{4}^{0} , A and B, and the quantities $|(A^2-B^2)^2/A|$ and $(A+2B)/3^2$ in gauss as a function of temperature.

т ([°] К)	b ₂ ⁰	b4	A	В	$\frac{\left(\mathbf{A}^2-\mathbf{B}^2\right)^{\frac{1}{2}}}{\mathbf{A}}$	<u>(A+2B)</u> 3
84.0000	-16.0000	-2+2500	-93•1100	-91.3300	•1945	-91.9233
99.0000	-15.8500	-2.3000	-93.1500	-91.3500	•1956	-91.9500
106.0000	-15.7500	-2.3000	-93•1000	-91.3500	•1929	-91.9333
121.0000	-15.5000	-2.3000	-93+1000	-91.3500	•1929	-91 •9333
133.0000	-15.1000	-2.3000	-93.0500	-91.3000	•1930	-91 •8833
153.0000	-13.8000	-2.3000	-93.0000	-91.1000	•2011	-91.7333
171.0000	-13.0000	-2.3000	-92.8500	-91.0500	•1959	-91 •6500
196.0000	-12.2200	-2.2800	-92•7200	-90.9000	•1971	-91 • 5066
225.0000	-10.6000	-2•3000	-92.6000	-90.6500	•2041	-91 • 3000
248.0000	-9.6000	-2.2500	-92•5 500	-90.6500	•2015	-91 • 2833
276.0000	-8.1000	-2•2600	-92•2500	-90.4000	•1992	-91.0166
290.0000	-6.6000	-2.5400	-92.0500	-90.3000	•1940	-90.8833
322.0000	-5.0000	-2.1000	-91.6000	-90.2100	•1735	-90.6733
344.0000	-4.4000	-2•2600	-91 • 4800	-90.1000	•1730	-90•5600
377.0000	9000	-2.2000	-91.3000	-89.8900	•1750	-90.3600

APPENDIX B (continued)

401.0000	-1.6000	-2.1000	-91.2000	-89.7000	•1806	-90.2000
453.0000	1.0000	-2.2000	-90•6700	-89•4300	•1648	-89.8433
493.0000	2.0000	-2.4000	-90•5500	-89•1500	•1751	-89•6166
516.0000	4.0000	-2.5000	-90•4000	-88•9200	•1802	-89.4133
544.00 00	4.8000	-2.5000	-90.1000	-88.7400	•1730	-89+1933
550.0000	6.2000	-2.2000	-90•1500	-88.8000	•1724	-89•2500
579.00 00	7.0000	-2•2000	-89•9000	-88.7100	•1621	-89 • 1066
605.0000	7.8000	-2.5000	-89•8500	-88.6000	•1662	-89.0166
636.0000	10.0000	-2.2600	-89•7000	-88.3500	•1728	-88.8000
680.0000	12.0000	-2.3000	-89•4000	-88.2200	•1619	-88.6133
705.0000	12.6000	-2.2800	-89•3300	-87•9100	•1775	-88 • 3 833
735.0000	13.0000	-2.3000	-89.1000	-87.7500	•1734	-88.2000
750.0000	13.8000	-2.3000	-89.0000	-87.6700	•1722	-88.1133
761.0000	15.2000	-2.3000	-89.0000	-87.6700	•1722	-88.1133
781.0000	16.9000	-2.2800	-88•8000	-87.7000	•1569	-88.0666

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VITA AUCTORIS

On August 1, 1942, I was born in Redhill, Surrey, England. In 1946, I emigrated to Essex county in Ontario where I graduated from S.S. No. 1 Public School, Maidstone, in 1956 and Essex District High School, in 1961. I received my university education at the University of Windsor, graduating with B.Sc. in 1965 and with M.Sc. in 1966.